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(54) Title: MICROEMULSION ALL PURPOSE LIQUID CLEANING COMPOSITIONS		
(57) Abstract An improvement is described in the microemulsion compositions which is especially effective in the removal of oily and greasy soil, contains an anionic detergent, a saturated fatty acid, an unsaturated fatty acid, a solubilizing agent, a perfume having a pine like odor, an alkali metal hydroxide and water.		

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MICROEMULSION ALL PURPOSE LIQUID CLEANING COMPOSITIONS

Field of the Invention

This invention relates to an improved all-purpose liquid cleaner in the form of a
5 microemulsion designed in particular for cleaning hard surfaces and which is effective
in removing grease soil and/or bath soil and in leaving unrinsed surfaces with a shiny
appearance.

Background of the Invention

In recent years all-purpose liquid detergents have become widely accepted for
10 cleaning hard surfaces, e.g., painted woodwork and panels, tiled walls, wash bowls,
bathtubs, linoleum or tile floors, washable wall paper, etc.. Such all-purpose liquids
comprise clear and opaque aqueous mixtures of water-soluble synthetic organic
detergents and water-soluble detergent builder salts. In order to achieve comparable
cleaning efficiency with granular or powdered all-purpose cleaning compositions, use of
15 water-soluble inorganic phosphate builder salts was favored in the prior art all-purpose
liquids. For example, such early phosphate-containing compositions are described in
U.S. Patent Nos. 2,560,839; 3,234,138; 3,350,319; and British Patent No. 1,223,739.

In view of the environmentalist's efforts to reduce phosphate levels in ground
water, improved all-purpose liquids containing reduced concentrations of inorganic
20 phosphate builder salts or non-phosphate builder salts have appeared. A particularly
useful self-opacified liquid of the latter type is described in U.S. Patent No. 4,244,840.

However, these prior art all-purpose liquid detergents containing detergent
builder salts or other equivalent tend to leave films, spots or streaks on cleaned
unrinsed surfaces, particularly shiny surfaces. Thus, such liquids require thorough
25 rinsing of the cleaned surfaces which is a time-consuming chore for the user.

In order to overcome the foregoing disadvantage of the prior art all-purpose
liquid, U.S. Patent No. 4,017,409 teaches that a mixture of paraffin sulfonate and a
reduced concentration of inorganic phosphate builder salt should be employed.

However, such compositions are not completely acceptable from an environmental

point of view based upon the phosphate content. On the other hand, another alternative to achieving phosphate-free all-purpose liquids has been to use a major proportion of a mixture of anionic and nonionic detergents with minor amounts of glycol ether solvent and organic amine as shown in U.S. Patent NO. 3,935,130. Again, this approach has not been completely satisfactory and the high levels of organic detergents necessary to achieve cleaning cause foaming which, in turn, leads to the need for thorough rinsing which has been found to be undesirable to today's consumers.

Another approach to formulating hard surfaced or all-purpose liquid detergent composition where product homogeneity and clarity are important considerations involves the formation of oil-in-water (o/w) microemulsions which contain one or more surface-active detergent compounds, a water-immiscible solvent (typically a hydrocarbon solvent), water and a "cosurfactant" compound which provides product stability. By definition, an o/w microemulsion is a spontaneously forming colloidal dispersion of "oil" phase particles having a particle size in the range of 25 Å to 800 Å in a continuous aqueous phase.

In view of the extremely fine particle size of the dispersed oil phase particles, microemulsions are transparent to light and are clear and usually highly stable against phase separation.

Patent disclosures relating to use of grease-removal solvents in o/w microemulsions include, for example, European Patent Applications EP 0137615 and EP 0137616 - Herbots et al; European Patent Application EP 0160762 - Johnston et al; and U.S. Patent No. 4,561,991 - Herbots et al. Each of these patent disclosures also teaches using at least 5% by weight of grease-removal solvent.

It also is known from British Patent Application GB 2144763A to Herbots et al, published March 13, 1985, that magnesium salts enhance grease-removal performance of organic grease-removal solvents, such as the terpenes, in o/w microemulsion liquid detergent compositions. The compositions of this invention described by Herbots et al. require at least 5% of the mixture of grease-removal solvent

and magnesium salt and preferably at least 5% of solvent (which may be a mixture of water-immiscible non-polar solvent with a sparingly soluble slightly polar solvent) and at least 0.1% magnesium salt.

However, since the amount of water immiscible and sparingly soluble components which can be present in an o/w microemulsion, with low total active ingredients without impairing the stability of the microemulsion is rather limited (for example, up to 18% by weight of the aqueous phase), the presence of such high quantities of grease-removal solvent tend to reduce the total amount of greasy or oily soils which can be taken up by and into the microemulsion without causing phase separation.

The following representative prior art patents also relate to liquid detergent cleaning compositions in the form of o/w microemulsions: U.S. Patents Nos. 4,472,291 - Rosario; 4,540,448 - Gauteer et al; 3,723,330 - Sheflin; etc.

Liquid detergent compositions which include terpenes, such as d-limonene, or other grease-removal solvent, although not disclosed to be in the form of o/w microemulsions, are the subject matter of the following representative patent documents: European Patent Application 0080749; British Patent Specification 1,603,047; 4,414,128; and 4,540,505. For example, U.S. Patent No. 4,414,128 broadly discloses an aqueous liquid detergent composition characterized by, by weight:

- (a) from 1% to 20% of a synthetic anionic, nonionic, amphoteric or zwitterionic surfactant or mixture thereof;
- (b) from 0.5% to 10% of a mono- or sesquiterpene or mixture thereof, at a weight ratio of (a):(b) lying in the range of 5:1 to 1:3; and
- (c) from 0.5% to 10% of a polar solvent having a solubility in water at 15°C in the range of from 0.2% to 10%. Other ingredients present in the formulations disclosed in this patent include from 0.05% to 2% by weight of an alkali metal, ammonium or alkanolammonium soap of a C₁₃-C₂₄ fatty acid; a calcium sequestrant from 0.5% to 13% by weight; non-aqueous solvent, e.g., alcohols and glycol ethers, up to 10% by weight; and hydrotropes, e.g., urea, ethanolamines, salts of lower alkylaryl sulfonates,

up to 10% by weight. All of the formulations shown in the Examples of this patent include relatively large amounts of detergent builder salts which are detrimental to surface shine.

Furthermore, the present inventors have observed that in formulations
5 containing grease-removal assisting magnesium compounds, the addition of minor amounts of builder salts, such as alkali metal polyphosphates, alkali metal carbonates, nitrilotriacetic acid salts, and so on, tends to make it more difficult to form stable microemulsion systems.

U.S. Patent 5,082,584 discloses a microemulsion composition having an anionic
10 surfactant, a cosurfactant, nonionic surfactant, perfume and water.

Summary of the Invention

The present invention provides an improved, clear, liquid cleaning composition having improved interfacial tension which improves cleaning hard surface in the form of a microemulsion which is suitable for cleaning hard surfaces such as plastic, vitreous
15 and metal surfaces having a shiny finish. More particularly, the improved cleaning compositions exhibit good grease soil removal properties due to the improved interfacial tensions, when used in undiluted (neat) form and leave the cleaned surfaces shiny without the need of or requiring only minimal additional rinsing or wiping. The latter characteristic is evidenced by little or no visible residues on the unrinsed cleaned
20 surfaces and, accordingly, overcomes one of the disadvantages of prior art products. The instant compositions exhibit a blooming effect, when the composition is added to water in that the formed solution is cloudy.

In one aspect, the invention generally provides a stable, clear all-purpose, hard surface cleaning composition especially effective in the removal of oily and greasy oil,
25 which is in the form of a substantially dilute oil-in-water microemulsion having an aqueous phase and an oil phase. The dilute o/w microemulsion includes, on a weight basis:

from 0.25% to 7%, more preferably 0.5% to 5.0% of an anionic surfactant;

from 0.1% to 10%, more preferably 0.5% to 7%, of a water-mixable cosurfactant having either limited ability or substantially no ability to dissolve oily or greasy soil;

0.5% to 10%, more preferably 0.75% to 7%, of an unsaturated fatty acid;

0.5% to 10%, more preferably 0.75% to 7%, of a fatty acid;

5 0.5% to 20.0%, more preferably 1% to 8% of a perfume having a pine like odor wherein 5 to 40 wt. % of the perfume is an eucalyptus oil;

0.1% to 8%, more preferably 0.5% to 6% of a solubilizing agent;

0.1% to 5%, more preferably 0.5% to 4.5% of an alkali metal hydroxide;

10% to 85% of water, wherein the composition does not contain pine oil.

10 Detailed Description of the Invention

The present invention relates to a stable liquid crystal or microemulsion composition approximately by weight: 0.25% to 7% of an anionic surfactant, 0.1% to 5% of an alkali metal hydroxide, 0.1% to 10% of a cosurfactant, 0.5% to 20% of a perfume having a pine like odor, wherein the perfume contains 5 to 40 wt. % of an eucalyptus oil, insoluble hydrocarbon, 0.1% to 8% of a solubilizing agent, 0.5% to 10%
15 of a saturated fatty acid; 0.5% to 10% of an unsaturated fatty acid and the balance being water.

The perfume which is employed in the instant composition has a pine like odor and contains 5 to 40 wt. %, more preferably 10 to 25 wt. % of an eucalyptus oil and 1 to
20 20 wt. %, more preferably 3 to 15 wt. % of limonene. The perfume is present in the composition at a concentration of about 0.5 to 20 wt. %, more preferably 1 to 8 wt. %.

The water-soluble organic surfactant materials which are used in forming the ultimate o/w microemulsion compositions of this invention are water-soluble, non-soap, anionic surfactants.

25 Regarding the anionic surfactant present in the o/w microemulsions any of the conventionally used water-soluble anionic surfactants or mixtures of said anionic detergents and anionic detergents can be used in this invention. As used herein the term "anionic surfactant" is intended to refer to the class of anionic and mixed anionic-nonionic surfactants providing deterative action.

alkyl group and are obtained by

ulfonate surfactants are the paraffin
3 to 17, carbon atoms. Primary paraffin
alpha olefins and bisulfites and paraffin
uted along the paraffin chain are shown in
260,744; 3,372,188; and German Patent

ate surfactants are the C₈-C₁₈ alkyl sulfate
the C₈-C₁₈ alkyl ether polyethenoxy sulfate
M wherein n is 1 to 12, preferably 1 to 5,
the group consisting of sodium, potassium,
triethanol ammonium ions. The alkyl
cohols obtained by reducing glycerides of
neutralizing the resultant product. On the
fates are obtained by sulfating the
a C₈-C₁₈ alkanol and neutralizing the
obtained by sulfating the alcohols obtained
w or mixtures thereof and neutralizing the
alkyl ether polyethenoxy sulfates are
luct of ethylene oxide with a C₈-C₁₈ alkanol
alkyl ether polyethenoxy sulfates differ from
lene oxide reacted with one mole of alkanol.
ether polyethenoxy sulfates contain 10 to 16
henoxy sulfates containing from 2 to 6
o are suitable for use in the inventive
pared by reacting an alkyl phenol with 2 to

nic surfactants include those surface-
an organic hydrophobic group containing
ly 10 to 18 carbon atoms in their molecular
group selected from the group of sulfonate,
-soluble surfactant. Usually, the
C₈-C₂₂ alkyl, alkyl or acyl group. Such
r-soluble salts and the salt-forming cation
of sodium, potassium, ammonium,
olammonium, with the sodium,
ing preferred.

ic surfactants are the well known higher
as the higher alkyl benzene sulfonates
higher alkyl group in a straight or
ates and C₈-C₁₅ alkyl phenol sulfonates.
zene sulfonate having a high content of
rdingly low content (well below 50%) of 2-
benzene ring is preferably attached in
, 6 or 7) position of the alkyl group and
ne ring is attached in the 2 or 1 position is
aterials are set forth in U.S. Patent

the olefin sulfonates, including long-chain
sulfonates or mixtures of alkene
ese olefin sulfonate detergents may be
of sulfur trioxide (SO₃) with long-chain
carbon atoms and having the formula
of 6 to 23 carbons and R₁ is an alkyl group
ure of sultones and alkene sulfonic acids
o sulfonates. Preferred olefin sulfonates

The preferred polyvalent metals are the di- and tri-valent metals of Groups IIA, IIB and IIIB, such as magnesium, calcium, aluminum and zinc, although other polyvalent metals, including those of Groups IIIA, IVA, VA, IB, IVB, VB, VIB, VIIB and VIII of the Periodic Table of the Elements can also be used. Specific examples of such other polyvalent metals include Ti, Zr, V, Nb, Mn, Fe, Co, Ni, Cd, Sn, Sb, Bi, etc. Generally, the metals may be present in the divalent to pentavalent state. Preferably the metal salts are used in their higher oxidation states. Naturally, for use in automatic dishwashers, as well as any other applications where the invention composition will or may come in contact with articles used for the handling, storage or serving of food products or which otherwise may come into contact with or be consumed by people or animals, the metal salt should be selected by taking into consideration the toxicity of the metal. For this purpose, the alkali metal and calcium and magnesium salts are especially higher preferred as generally safe food additives.

Generally, however, amounts of the saturated fatty acid or saturated fatty acid salt is agents in the range of from 0.5 to 10 wt. %, more preferably 0.75 to 7 wt. %.

The preferred long chain unsaturated fatty acids of the instant invention have about 8 to about 24 carbon atoms, more preferably about 10 to about 20 carbon atoms. A preferred unsaturated fatty acid mixture is a refined tall oil fatty acid. A typical tall oil fatty acid contains mono unsaturated C₁₆₋₁₈ fatty acid; a C₁₈ diene unsaturated fatty acid; a C₁₆₋₁₈ triene unsaturated fatty acid; and a C₁₆₋₁₈ saturated fatty acid. Other unsaturated fatty acids that are usable in the instant compositions are unsaturated vegetable oil fatty acids, including soy, peanut, corn, cottonseed, linseed and refined oleic fatty acids, and fatty acids consisting predominantly of C₁₈ (average) unsaturated fatty acids and mixtures thereof. The unsaturated fatty acid reacts in situ with the alkali metal hydroxide to form the alkali metal salt of the unsaturated fatty acid. The concentration of the unsaturated fatty acid is about 0.5 to about 10 wt %, more preferably about 0.75 to about 7 wt %. The alkali metal hydroxide present in the composition is preferably potassium hydroxide and is present in the composition at a concentration of about 0.1 to about 5 wt %, more preferably about 0.5 to about 4.5

wt %. The potassium hydroxide reacts in situ with both the saturated and unsaturated fatty acid in the composition to form the potassium salts of the saturated fatty acid.

The solubilizing agents are water soluble hydrotropic salts of xylene or cumene sulfonate include sodium, potassium, ammonium and mono-, di- and triethanolammonium salts of xylene or cumene sulfonate or C₂-C₄ alkanols or dealkanols such as isopropanol. While the aqueous medium is primarily water, preferably said solubilizing agents are included in order to control the viscosity of the liquid composition and to control low temperature cloud clear properties. Usually, it is desirable to maintain clarity to a temperature in the range of 5°C to 10°C. Therefore, the proportion of solubilizer generally will be from 1%-15%, preferably 2%-12%, most preferably 2%-8%, by weight of the detergent composition with the proportion of ethanol, when present, being 5% of weight or less in order to provide a composition having a flash point above 46°C. The solubilizing ingredient can be a mixture of isopropanol or ethanol and either sodium xylene sulfonate or sodium cumene sulfonate or a mixture of said sulfonates or ethanol or isopropanol and urea.

The instant compositions contain about 0.1 wt. % to about 8 wt. %, more preferably about 0.5 wt. % to about 6 wt. %, of at least one solubilizing agent which is a C₂-5 mono, dihydroxy or polyhydroxy alkanols such as ethanol, isopropanol, glycerol ethylene glycol, diethylene glycol and propylene glycol and mixtures thereof. The solubilizing agents are included in order to control low temperature cloud clear properties. Urea can be optionally employed in the instant composition as a supplemental solubilizing agent at a concentration of 0 to about 10 wt. %, more preferably about 0.5 wt. % to about 8 wt. %.

The cosurfactant may play an essential role in the formation of the microemulsion. Very briefly, in the absence of the cosurfactant the water, detergent(s) and hydrocarbon (e.g., perfume) will, when mixed in appropriate proportions form either a micellar solution (low concentration) or form an oil-in-water emulsion in the first aspect of the invention. With the cosurfactant added to this system, the interfacial tension at the interface between the emulsion droplets and aqueous phase is reduced

to a very low value. This reduction of the interfacial tension results in spontaneous break-up of the emulsion droplets to consecutively smaller aggregates until the state of a transparent colloidal sized emulsion. e.g., a microemulsion, is formed. In the state of a microemulsion, thermodynamic factors come into balance with varying degrees of stability related to the total free energy of the microemulsion. Some of the thermodynamic factors involved in determining the total free energy of the system are (1) particle-particle potential; (2) interfacial tension or free energy (stretching and bending); (3) droplet dispersion entropy; and (4) chemical potential changes upon formation. A thermodynamically stable system is achieved when (2) interfacial tension or free energy is minimized and (3) droplet dispersion entropy is maximized.

Thus, the role of cosurfactant in formation of a stable o/w microemulsion is to (a) decrease interfacial tension (2); and (b) modify the microemulsion structure and increase the number of possible configurations (3). Also, the cosurfactant will (c) decrease the rigidity. Generally, an increase in cosurfactant concentration results in a wider temperature range of the stability of the product.

The major class of compounds found to provide highly suitable cosurfactants for the microemulsion over temperature ranges extending from 5°C to 43°C for instance are water-soluble polyethylene glycols having a molecular weight of 150 to 1000, polypropylene glycol of the formula $\text{HO}(\text{CH}_2\text{CHCH}_2\text{O})_n\text{H}$ wherein n is a number from 2 to 18, mixtures of polyethylene glycol and polypropylene glycol (Synalox) and mono and di C_1 - C_6 alkyl ethers and esters of ethylene glycol and propylene glycol having the structural formulas $\text{R}(\text{X})_n\text{OH}$, $\text{R}_1(\text{X})_n\text{OH}$, $\text{R}(\text{X})_n\text{OR}$ and $\text{R}_1(\text{X})_n\text{OR}_1$ wherein R is C_1 - C_6 alkyl group, R_1 is C_2 - C_4 acyl group, X is $(\text{OCH}_2\text{CH}_2)$ or $(\text{OCH}_2(\text{CH}_3)\text{CH})$ and n is a number from 1 to 4, diethylene glycol, triethylene glycol, an alkyl lactate, wherein the alkyl group has 1 to 6 carbon atoms, 1-methoxy-2-propanol, 1-methoxy-3-propanol, and 1-methoxy 2-, 3- or 4-butanol.

Representative members of the polypropylene glycol include dipropylene glycol and polypropylene glycol having a molecular weight of 150 to 1000, e.g., polypropylene glycol 400. Other satisfactory glycol ethers are ethylene glycol monobutyl ether (butyl

cellosolve), diethylene glycol monobutyl ether (butyl carbitol), triethylene glycol monobutyl ether, mono, di, tri propylene glycol monobutyl ether, tetraethylene glycol monobutyl ether, mono, di, tripropylene glycol monomethyl ether, propylene glycol monomethyl ether, ethylene glycol monohexyl ether, diethylene glycol monohexyl ether, 5 propylene glycol tertiary butyl ether, ethylene glycol monoethyl ether, ethylene glycol monomethyl ether, ethylene glycol monopropyl ether, ethylene glycol monopentyl ether, diethylene glycol monomethyl ether, diethylene glycol monoethyl ether, diethylene glycol monopropyl ether, diethylene glycol monopentyl ether, triethylene glycol monomethyl ether, triethylene glycol monoethyl ether, triethylene glycol monopropyl 10 ether, triethylene glycol monopentyl ether, triethylene glycol monohexyl ether, mono, di, tripropylene glycol monoethyl ether, mono, di tripropylene glycol monopropyl ether, mono, di, tripropylene glycol monopentyl ether, mono, di, tripropylene glycol monohexyl ether, mono, di, tributylene glycol mono methyl ether, mono, di, tributylene glycol monoethyl ether, mono, di, tributylene glycol monopropyl ether, mono, di, tributylene 15 glycol monobutyl ether, mono, di, tributylene glycol monopentyl ether and mono, di, tributylene glycol monohexyl ether, ethylene glycol monoacetate and dipropylene glycol propionate. These glycol type cosurfactants are at a concentration of about 0.1 to about 10 weight %, more preferably about 0.5 weight % to about 7 weight %.

While all of the aforementioned glycol ether compounds provide the described 20 stability, the most preferred cosurfactant compounds of each type, on the basis of cost and cosmetic appearance (particularly odor), are dipropylene glycol monomethyl ether and diethylene glycol monobutyl ether. Other suitable water soluble cosurfactants are water soluble esters such as ethyl lactate and water soluble carbohydrates such as butyl glycosides.

25 The amount of cosurfactant required to stabilize the microemulsion compositions will, of course, depend on such factors as the surface tension characteristics of the cosurfactant, the type and amounts of the primary surfactants and water insoluble hydrocarbon, and the type and amounts of any other additional ingredients which may be present in the composition and which have an influence on the thermodynamic

factors enumerated above. Generally, amounts of cosurfactant in the range of from 0.1% to 10%, preferably from about 0.5 wt. % to 7 wt. % provide stable dilute o/w microemulsions for the above-described levels of primary surfactants and water insoluble hydrocarbon and any other additional ingredients as described below.

5 The combination of solubilizing agents and cosurfactants such as isopropanol and diethylene glycol monobutyl ether and perfume which has a pine like odor which are used in the invention tend to compatibilize the water insoluble within the aqueous dispersion such that clear stable dispersions are formed which will not separate or cloud up after period of standing. Also the solvents appear to act synergistically such
10 that lesser amounts of the combination is required to achieve a stable dispersion than would be the case, if each type of solvent was used alone as the sole solvent source. This phenomena allows for a significantly lower VOC content in the composition which is more enviromentally preferred. The isopropanol and diethylenglycol monobutyl ether may be mixed at a respective weight ratio of from about 1:2 to 2:1, more preferably at a
15 ratio of 0.75:1 to 1.25:1 and most preferably at a ratio of 0.8:1 to 1:1 of alcohol and glycol ether respectively. A preferred solvent combination is isopropanol and diethylenglycol monobutyl ether.

 The final essential ingredient in the inventive microemulsion compositions having improved interfacial tension properties is water. The proportion of water in the
20 microemulsion compositions generally is in the range of 20% to 97%, preferably 70% to 97% by weight of the usual diluted o/w microemulsion composition.

 In addition to the above-described essential ingredients required for the formation of the liquid crystal composition or the microemulsion composition, the compositions of this invention may often and preferably do contain one or more
25 additional ingredients which serve to improve overall product performance.

 One such ingredient is an inorganic or organic salt of oxide of a multivalent metal cation, particularly Mg^{++} . The metal salt or oxide provides several benefits including improved cleaning performance in dilute usage, particularly in soft water areas, and minimized amounts of perfume required to obtain the microemulsion state. Magnesium

sulfate, either anhydrous or hydrated (e.g., heptahydrate), is especially preferred as the magnesium salt. Good results also have been obtained with magnesium oxide, magnesium chloride, magnesium acetate, magnesium propionate and magnesium hydroxide. These magnesium salts can be used with formulations at neutral or acidic pH since magnesium hydroxide will not precipitate at these pH levels.

Although magnesium is the preferred multivalent metal from which the salts (inclusive of the oxide and hydroxide) are formed, other polyvalent metal ions also can be used provided that their salts are nontoxic and are soluble in the aqueous phase of the system at the desired pH level. Thus, depending on such factors as the pH of the system, the nature of the primary surfactants and cosurfactant, and so on, as well as the availability and cost factors, other suitable polyvalent metal ions include aluminum, copper, nickel, iron, calcium, etc. It should be noted, for example, that with the preferred paraffin sulfonate anionic detergent calcium salts will precipitate and should not be used. It has also been found that the aluminum salts work best at pH below 5 or when a low level, for example 1 weight percent, of citric acid is added to the composition which is designed to have a neutral pH. Alternatively, the aluminum salt can be directly added as the citrate in such case. As the salt, the same general classes of anions as mentioned for the magnesium salts can be used, such as halide (e.g., bromide, chloride), sulfate, nitrate, hydroxide, oxide, acetate, propionate, etc.

Preferably, in the dilute compositions the metal compound is added to the composition in an amount sufficient to provide at least a stoichiometric equivalence between the anionic surfactant and the multivalent metal cation. For example, for each gram-ion of Mg^{++} there will be 2 gram moles of paraffin sulfonate, alkylbenzene sulfonate, etc., while for each gram-ion of Al^{3+} there will be 3 gram moles of anionic surfactant. Thus, the proportion of the multivalent salt generally will be selected so that one equivalent of compound will neutralize from 0.1 to 1.5 equivalents, preferably 0.9 to 1.4 equivalents, of the acid form of the anionic surfactant.

At higher concentrations of anionic surfactant, the amount of multivalent salt will be in range of 0.5 to 1 equivalents per equivalent of anionic surfactant.

The all-purpose liquid cleaning composition of this invention may, if desired, also contain other components either to provide additional effect or to make the product more attractive to the consumer. The following are mentioned by way of example: Colors or dyes in amounts up to 0.5% by weight; bactericides in amounts up to 1% by weight; preservatives or antioxidizing agents, such as formalin, 5-bromo-5-nitro-dioxan-1,3; 5-chloro-2-methyl-4-isothiazolin-3-one, 2,6-di-tert.butyl-p-cresol, etc., in amounts up to 2% by weight; and pH adjusting agents, such as sulfuric acid or sodium hydroxide, as needed. Furthermore, if opaque compositions are desired, up to 4% by weight of an opacifier may be added.

The instant compositions of the instant invention explicitly exclude zwitterionic surfactant such as betaines because these zwitterionic surfactants are extremely high foaming which, if used in the instant composition, would cause the instant compositions to have to high a foam profile and that too much foam would leave residue on the surface being cleaned.

In final form, the all-purpose liquids are low foaming, clear oil-in-water microemulsions or liquid crystal compositions and exhibit stability at reduced and increased temperatures. More specifically, such compositions remain clear and stable in the range of 5°C to 50°C, especially 10°C to 43°C. Such compositions exhibit a pH in the acid or neutral range depending on intended end use. The liquid microemulsion compositions are readily pourable and exhibit a viscosity in the range of 6 to 60 milliPascal . second (mPas.) as measured at 25°C. with a Brookfield RVT Viscometer using a #1 spindle rotating at 20 RPM. Preferably, the viscosity is maintained in the range of 10 to 40 mPas.

The compositions are directly ready for use or can be diluted as desired and in either case no or only minimal rinsing is required and substantially no residue or streaks are left behind. Furthermore, because the compositions are free of detergent builders such as alkali metal polyphosphates they are environmentally acceptable and provide a better "shine" on cleaned hard surfaces.

When intended for use in the neat form, the liquid compositions can be packaged under pressure in an aerosol container or in a pump-type sprayer for the so-called spray-and-wipe type of application.

Because the compositions as prepared are aqueous liquid formulations and since no particular mixing is required to form the o/w microemulsion, the compositions are easily prepared simply by combining all the ingredients in a suitable vessel or container. The order of mixing the ingredients is not particularly important and generally the various ingredients can be added sequentially or all at once or in the form of aqueous solutions of each or all of the primary detergents and cosurfactants can be separately prepared and combined with each other and with the perfume. It was seen that making a premix of the solubilizing agent, cosurfactant and perfume (isopropanol, diethylene glycol monobutyl ether and fragrance) considerably reduces the mixing time, helping to achieve the emulsion, and could help in reducing the amount of solubilizer and/or cosurfactant needed in order to have a clear stable product. The magnesium salt, or other multivalent metal compound, when present, can be added as an aqueous solution thereof or can be added directly. It is not necessary to use elevated temperatures in the formation step and room temperature is sufficient.

The instant microemulsion formulas explicitly exclude alkali metal silicates and alkali meta builders such as alkali metal polyphosphates, alkali metal carbonates, alkali metal phosphonates and alkali metal citrates because these materials, if used in the instant composition, would cause the composition to have a high pH as well as leaving residue on the surface being cleaned.

The following examples illustrate liquid cleaning compositions of the described invention. Unless otherwise specified, all percentages are by weight. The exemplified compositions are illustrative only and do not limit the scope of the invention. Unless otherwise specified, the proportions in the examples and elsewhere in the specification are by weight.

Example 1

The following compositions in wt. % were prepared:

	A	B	C	D	E	F	G	H
Sodium dodecyl benzene sulfonate C13-C17 sulfonate (48% A.I)	2.17	2.17	2.17	2.17	2.17	2.17	2.17	2.17
Stearic acid	2.0	0.0	0.0	0.0	0.0	2.0	2.0	0
Coconut fatty acid	2.3	4.3	4.3	4.3	4.3	2.3	2.3	9.6
Isopropanol	2.0	2.0	2.0	0.0	2.5	2.0	4.5	2.0
Diethylene glycol monobutyl ether	2.5	2.5	2	4.5	2	2.5	0	2.5
Potassium hydroxide	2.7	2.7	2.7	2.7	2.7	2.7	2.7	5.8
Perfume (a)	5.0	10	10	10	10	10	10	20
Water	Bal.	Bal	Bal	Bal	Bal	Bal	Bal	Bal
pH	12	12- 13	uns	uns	12- 13	12- 13	uns	12- 13
Degreasing test								
Neat (b)	10	9	--	--	12	9	--	6
Dilute (b)	28	27	--	--	32	28	--	22
Residue	STD	EQ			EQ	EQ		EQ
Foam in hard Water	STD	EQ			EQ	EQ		EQ
Cloud point (LC _i)	>8	>8			12	>8		>8

- 5 *uns= Unstable dispersion, separated or clouded immediately or on standing up to 1 day

(a) contains 5 to 15 wt. % of limonene and 12 to 22 wt. % of eucalyptus oil and the perfume has a pine like odor.

(b) the lower the number of strokes, the better the degreasing performance.

What Is Claimed Is

1. A microemulsion composition comprising:
 - (a) 0.25 wt. % to 7 wt. % of an anionic surfactant;
 - (b) 0.1 wt. % to 10 wt. % of a cosurfactant;
 - 5 (c) 0.5 wt. % to 20 wt. % of a perfume having a pine like odor;
 - (d) 0.5 wt. % to 10 wt. % of an unsaturated fatty acid;
 - (e) 0.5 wt. % to 10 wt. % of a saturated fatty acid;
 - (f) 0.1 wt. % to 5 wt. % of an alkali metal hydroxide;
 - (g) 0.1 wt. % to 8 wt. % of a solubilizing agent; and
 - 10 (h) the balance being water.
2. The cleaning composition of Claim 1 which further contains a salt of a multivalent metal cation in an amount sufficient to provide from 0.5 to 1.5 equivalents of said cation per equivalent of said anionic detergent.
3. The cleaning composition of Claim 2 wherein the multivalent metal cation
15 is magnesium or aluminum.
4. The cleaning composition of Claim 2; wherein said composition contains 0.9 to 1.4 equivalents of said cation per equivalent of anionic detergent.
5. The cleaning composition of Claim 3 wherein said multivalent salt is magnesium oxide, magnesium chloride or magnesium sulfate.
- 20 6. The cleaning composition of Claim 1 wherein said saturated fatty acid has 8 to 22 carbon atoms.
7. The cleaning composition of Claim 1 wherein said unsaturated fatty acid has 8 to 24 carbon atoms.
8. The cleaning composition of Claim 1 wherein the cosurfactant is a water
25 soluble glycol ether.
9. The cleaning composition of Claim 8 wherein the glycol ether is selected from the group consisting of ethylene glycol monobutyl ether, diethylene glycol monobutyl ether, triethylene glycol monobutyl ether, polypropylene glycol having an

INTERNATIONAL SEARCH REPORT

onal Application No

/US 98/09809

A. CLASSIFICATION OF SUBJECT MATTER

IPC 6 C11D17/00 C11D1/37 C11D3/18 C11D3/20 C11D3/04
//C11D1:04,C11D1:14,C11D1:22

According to International Patent Classification (IPC) or to both national classification and IPC

B. FIELDS SEARCHED

Minimum documentation searched (classification system followed by classification symbols)

IPC 6 C11D

Documentation searched other than minimum documentation to the extent that such documents are included in the fields searched

Electronic data base consulted during the international search (name of data base and, where practical, search terms used)

C. DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No
A	EP 0 620 271 A (COLGATE-PALMOLIVE CO.) 19 October 1994 see page 2, line 56 - page 4, line 36 see claims 1-15	1-13
A	EP 0 137 616 A (THE PROCTER & GAMBLE CO.) 17 April 1985 see page 5 - page 6 see page 9; claims 1-9	1-7, 12, 13
A	US 5 082 584 A (LOTH MYRIAM ET AL.) 21 January 1992 cited in the application see page 3, line 14 - page 11, line 32 see page 8, line 9 - line 33	1-13
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☒ Further documents are listed in the continuation of box C.

☒ Patent family members are listed in annex

* Special categories of cited documents :

"A" document defining the general state of the art which is not considered to be of particular relevance

"E" earlier document but published on or after the international filing date

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"O" document referring to an oral disclosure, use, exhibition or other means

"P" document published prior to the international filing date but later than the priority date claimed

"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention

"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step when the document is taken alone

"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art

"&" document member of the same patent family

Date of the actual completion of the international search

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Serbetsoglou, A

average molecular weight of from 200 to 1,000 and propylene glycol tert.butyl ether, mono, di, tri propylene glycol monobutyl ether.

10. The cleaning composition of Claim 9 wherein the glycol ether is ethylene glycol monobutyl ether or diethylene glycol monobutyl ether.

5 11. The cleaning composition of Claim 1 wherein said solubilizing agent is a C2-C4 alkanol.

12. The cleaning composition of Claim 1 wherein the anionic surfactant is a C9-C15 alkyl benzene sulfonate.

10 13. The cleaning composition of Claim 1, wherein said alkali metal hydroxide is potassium hydroxide.

14. The composition of Claim 1, wherein said perfume contains 5 to 40 wt. % of an eucalyptus oil.

INTERNATIONAL SEARCH REPORT

International Application No
PCT/US 98/09809

C.(Continuation) DOCUMENTS CONSIDERED TO BE RELEVANT

Category *	Citation of document, with indication, where appropriate, of the relevant passages	Relevant to claim No
A	<p>EP 0 316 726 A (COLGATE-PALMOLIVE CO.) 24 May 1989 see claims 1-8 see page 3, line 56 - page 6, line 34</p>	1-12

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US88/09809

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
EP 0620271 A	19-10-1994	AU 675500 B	06-02-1997
		AU 5948094 A	20-10-1994
		BR 9401481 A	18-10-1994
		CA 2117274 A	15-10-1994
		CN 1093743 A	19-10-1994
		HU 66866 A	30-01-1995
		JP 6330098 A	29-11-1994
		NZ 260307 A	28-03-1995
		PT 101489 A	04-05-1995
		US 5543073 A	06-08-1996
		US 5641742 A	24-06-1997
		ZA 9402545 A	13-10-1995
EP 0137616 A	17-04-1985	CA 1236372 A	10-05-1988
		DE 3469037 A	03-03-1988
		EG 16583 A	30-12-1991
		FI 843159 A, B,	12-02-1985
		JP 1907680 C	24-02-1995
		JP 6031420 B	27-04-1994
		JP 60106898 A	12-06-1985
US 5082584 A	21-01-1992	US 5075026 A	24-12-1991
		AT 957 U	26-08-1996
		AU 597367 B	31-05-1990
		AU 7313887 A	26-11-1987
		BE 1001742 A	27-02-1990
		BR 8702610 A	23-02-1988
		CA 1302194 A	02-06-1992
		CH 676854 A	15-03-1991
		DE 3716526 A	26-11-1987
		DK 258987 A	22-11-1987
		FR 2599046 A	27-11-1987
		GB 2190681 A, B	25-11-1987
		HK 60094 A	08-07-1994
		LU 86888 A	22-11-1988
		MX 169901 B	30-07-1993
		NL 8701215 A	16-12-1987
		PH 26507 A	07-08-1992
		PT 84886 B	08-02-1990
		SE 503826 C	16-09-1996

INTERNATIONAL SEARCH REPORT

Information on patent family members

International Application No

PCT/US 98/09809

Patent document cited in search report	Publication date	Patent family member(s)	Publication date
US 5082584 A		SE 8702084 A SG 20994 G TR 25009 A US 5076954 A	22-11-1987 10-06-1994 01-09-1992 31-12-1991
EP 0316726 A	24-05-1989	US 5076954 A AU 2506288 A CA 1337585 A DK 628488 A MX 169813 B PT 88977 A,B US 5108643 A	31-12-1991 18-05-1989 21-11-1995 13-05-1989 27-07-1993 30-11-1989 28-04-1992

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